



CHEMICAL REACTIONS AND THEIR BAL ANCING with LEGO!

THEIR BALANCING with LEGO!
Students
Class Date Group number
Try to build and balance these reactions using the LEGO bricks and the LEGO-periodic table projected on the interactive whiteboard.
CHEMICAL EQUATION 1
$Na + O_2 \longrightarrow Na_2O$

Build the molecules and put them in the right place:



Use the coefficient and write the **balanced equation**:

ELEMENTS	REACTANTS	PRODUCTS
Na		
0		

$$Fe + O_2 \longrightarrow Fe_2O_3$$

Build the molecules and put them in the right place:



Use the	coefficient a	and write the	balanced	${\bf equation}\colon$	

ELEMENTS	REACTANTS	PRODUCTS
Fe		
0		

$$Na + H_2O \longrightarrow NaOH + H_2$$

Build the molecules and put them in the right place:



Use the	coefficient	and write	the ba	alanced	equation:	

ELEMENTS	REACTANTS	PRODUCTS
Na		
Н		
0		



Build the molecules and put them in the right place:



Use	the	coeffic	ient	and	write	the	balanced	equa	ation:		

ELEMENT	REACTANTS	PRODUCTS
0		

$$Na + Cl_2 \longrightarrow NaCl$$

Build the molecules and put them in the right place:



Use the coeffice	cient and write t	the balanced	equation:	

ELEMENTS	REACTANTS	PRODUCTS
Na		
Cl		

$$Mg + O_2 \longrightarrow Mg_2O_3$$

Build the molecules and put them in the right place:



Use the co	efficient and	write the bal	anced equati	ion:	

ELEMENTS	REACTANTS	PRODUCTS
Mg		
0		

$$Fe_2O_3 + AI \longrightarrow Fe + AI_2O_3$$

Build the molecules and put them in the right place:



Use	the	coeffi	cient	and	write	the	bala	ance	d equ	ıatio	n:		

ELEMENTS	REACTANTS	PRODUCTS
Fe		
0		
Al		